

# Investigation of the percent contributions to 17 $\alpha$ -methyltestosterone depletion in a lab-scale *Salvinia*-based reactor

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## **Introduction**

- Background
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**Results &  
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Conclusion

## **Methodology**

- Materials
- Methods

## Emerging contaminants (ECs)

pesticides, pharmaceuticals and personal care products (PPCPs),  
**Endocrine Disruption Compounds (EDC)** etc.

EDCs substances

- An exogenous substance that interfere endocrine system and have adverse effect to living organism

Steroid Hormones

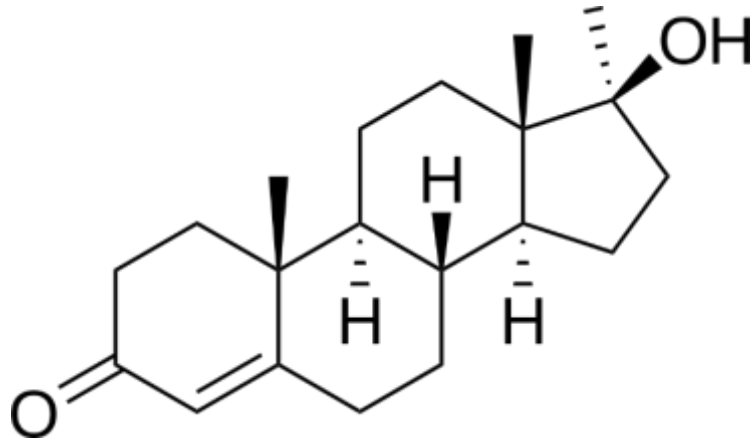
- Organic substances
- Toxic both natural and synthetic compounds

17 $\alpha$ - Methyltestosterone  
(MT)

- Synthetic androgen hormone
- Introduce to pharmaceutical and fish aquaculture application

# What is 17 $\alpha$ -Methyltestosterone (MT)?

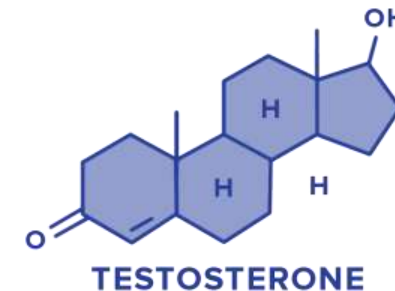
**Table 1.** Chemical and physical properties of 17 $\alpha$ -methyltestosterone [1]



**Figure 1.** 17 $\alpha$ -methyltestosterone  
(C<sub>20</sub>H<sub>30</sub>O<sub>2</sub>)

Molecular Formula	C <sub>20</sub> H <sub>30</sub> O <sub>2</sub>
Molecular Weight	302.45
Color/Form	White crystal powder
Odor	Odorless
Melting Point	161–166 °C
Solubilities	Methanol, ethanol, acetonitrile In water; 34 mg/L at 25 °C
Stabilities	In low pH level
pKa/log K <sub>ow</sub>	15.13/3.4

- Type: Anabolic steroid hormone
- Function: Hormonal growth promoter
- Close structural similarity to testosterone



# What are the applications of MT?

## 17 $\alpha$ - Methyltestosterone (MT)



- More muscles
- Stronger bones
- Improve mood
- Healthy heart and blood

- Reverse sex (all males)
- Increase production yield
- Fast growth rate
- Dosage: 60 mg 17MT/kg fish feed (Straus et al., 2013)

- Banned to used in sport activities
- Still used in fish aquaculture application

# What are the effects of using MT?



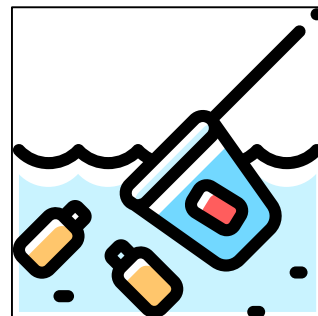
## **Interfere endocrine systems**

- Decreased testicular size
- Decrease sperm counts
- Increased aggressive behavior
- Breast enlargement (Combalbert et al., 2012)



## **Alter the endocrine in system of fish such as :**

- Transcriptome (Gao et al., 2015)
- Gonadal gland
- Embryo development (Rivero-Wendt et al., 2016)









## **MT enter the aquatic environment via :**

- Aquaculture discharge
- Domestic sewage
- Industrial discharge (Heidarimoghadam et al., 2016)

# MT Levels of which caused effects

**Table 2.** MT levels in water of which caused effects on the tested animals

MT levels (µg/L)	Tested animals	Effects	Reference
0.1-1	 Ramshorn snail	Affect the male sex organs	(Schulte-Oehlmann et al., 2004)
1.0	 <i>Gasterosteus aculeatus</i> L.	Cause intersex in both male and female of three-spined stickleback	(Hahlbeck et al., 2004)
0.01-0.5	 <i>Zoarces viviparus</i>	Decrease female characteristic in female eel spout for 10 days	(Korsgaard, 2006)
0.046	 medaka fish	Decrease the fecundity and fertility	(Kang et al., 2008)
4	 zebrafish	Inhibit vitellogenin (VTG) in zebrafish LC <sub>50</sub> = 10.09 mg MT/L within 96h	(Rivero-Wendt et al., 2016)
0.25-5	 <i>Biomphalaria glabrata</i>	Affect hatchling survival and growth rate of snail	(Kaur et al., 2016)

# Water and Wastewater Treatment Technology for MT

Technology	Pros	Cons
<ul style="list-style-type: none"> <li>biological treatment,</li> <li>conventional sewage treatment</li> </ul>	<ul style="list-style-type: none"> <li>Successfully remove soluble biodegradable organic contaminants from water</li> </ul>	<ul style="list-style-type: none"> <li>However, it cannot completely remove steroid hormones and MT</li> </ul>
<ul style="list-style-type: none"> <li>Advanced treatment technology (i.e. Advanced Oxidation Process)</li> </ul>	<ul style="list-style-type: none"> <li>Successfully remove organic contaminants from water</li> </ul>	<ul style="list-style-type: none"> <li>However, metabolites or intermediate products from complex reactions can lead to other questions regarding their toxicity to the environment</li> </ul>
<ul style="list-style-type: none"> <li>Natural biological treatment system incorporating aquatic plants</li> </ul>	<ul style="list-style-type: none"> <li>More effective and economical for the treatment of contaminated water with very low concentrations (parts per billion) of compounds such as surfactants, hormones, polar microcontaminants, and pharmaceutical compounds</li> </ul>	<ul style="list-style-type: none"> <li>However, performance in the removal of organic contaminants can be highly variable and is influenced by many factors, such as environmental factors and operational parameters.</li> <li>Little is known about the removal mechanisms involved</li> </ul>



# OBJECTIVES

- To the best of our knowledge, the potential use of *Salvinia* as a phytoremediator of MT has not been studied thus far.
- The results of our previous study showed that an active *Salvinia*-based reactor could be used for the removal of MT from contaminated water with a more than 90% removal efficiency.
- It was found that sorption was one of the key processes that accounted for the disappearance of MT from the water phase  
(Adnan & Thanasupsin, 2016).



To search for a scientific explanation regarding the removal efficiency of MT in a lab-scale *Salvinia*-based reactor.

# MATERIALS AND METHODS

## Determination of MT by HPLC-UV

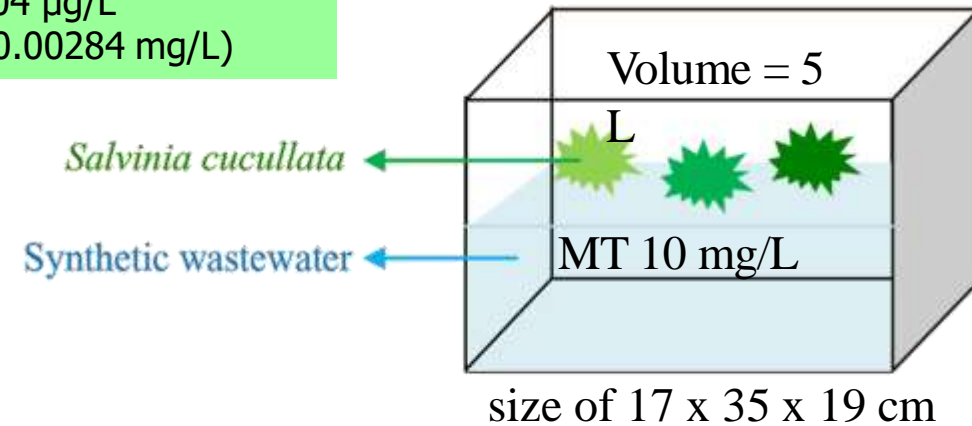
- The procedure for analyzing the water samples was modified from the protocol reported in (I. R. Barbosa et al., 2013),

## HPLC conditions

- The column used for the chromatographic separation of MT was a carbon 18 reversed-phase (RP-C 18) column (ACE, 5  $\mu\text{m}$  particle size, 250 x 4.6 mm).
- The injection of analytes was performed both automatically and manually manner at 25°C.
- The ratio and flow rate of the mobile phase used for MT detection (acetonitrile and DI water) were 70:30, v/v and 1 mL/min

LOD = 12.61  $\mu\text{g/L}$   
(0.01261 $\pm$ 0.00085 mg/L)

LOQ = 42.04  $\mu\text{g/L}$   
(0.04204 $\pm$ 0.00284 mg/L)



There are 50 g of Salvinia and 5 L of synthetic wastewater in the system

A composite sampling of 24 mL.

- Each sample was composed of samples collected from the top (8 mL), middle (8 mL), and bottom (8 mL) of the reactor.
- Water samples were collected in amber glass bottles and kept in the laboratory refrigerator

# Table 2 Details of the experimental trials.

- Seven batch experiments, including 4 controls and 3 trials, were performed in this research.
- Each experimental trial was conducted in triplicate.

Conditions	Ref.	Control 1	Control 2	Control 3	Control 4	Exp.5	Exp.6	Exp.7
MT 10 mg/L	(Adnan & Thanasupsin, 2016)	/	/	/	/	/	/	/
Plant nutrients	(Shi et al., 2010)	/	/	/	/	/	/	/
NaN <sub>3</sub> (1% w/w) <sup>a</sup>	-	/	/	-	/	/	/	-
Salvinia (50 g)	-	-	-	-	-	/ <sup>b</sup>	/	/
Top of the reactor covered	-	/	-	-	-	-	-	-
Natural light	-	- <sup>c</sup>	- <sup>c</sup>	/	/	/	/	/

<sup>a</sup> NaN<sub>3</sub> was used for microbial activity inhibition,

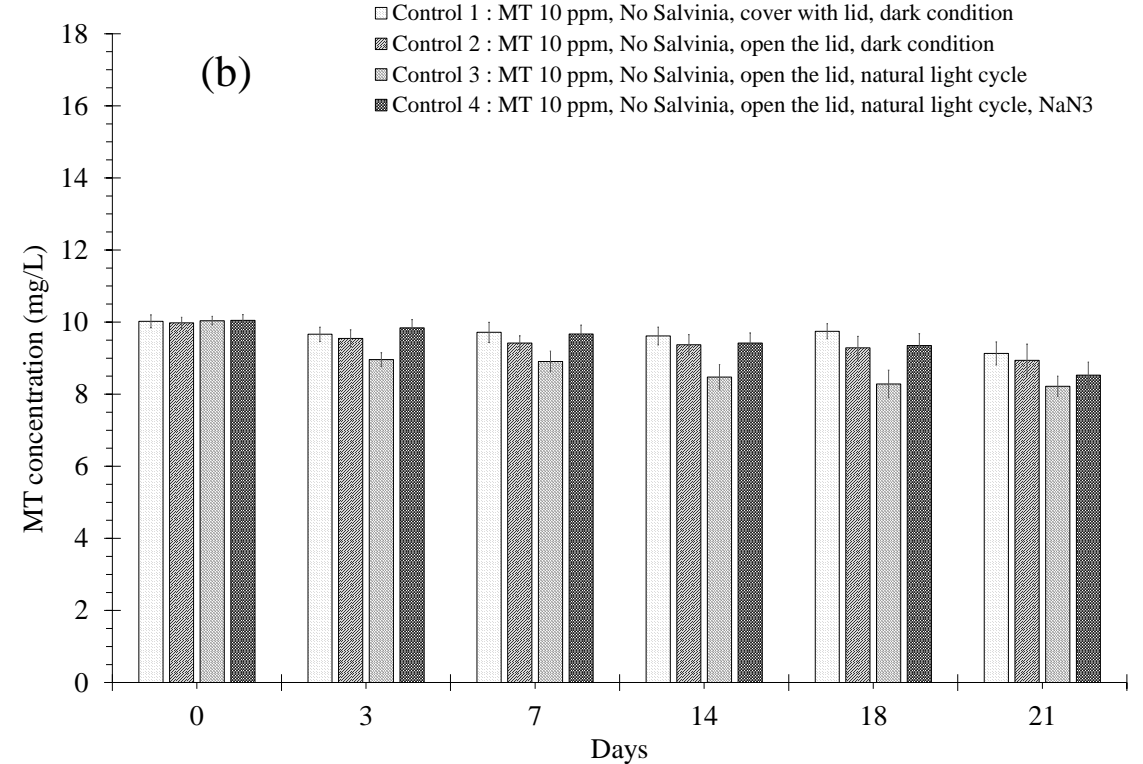
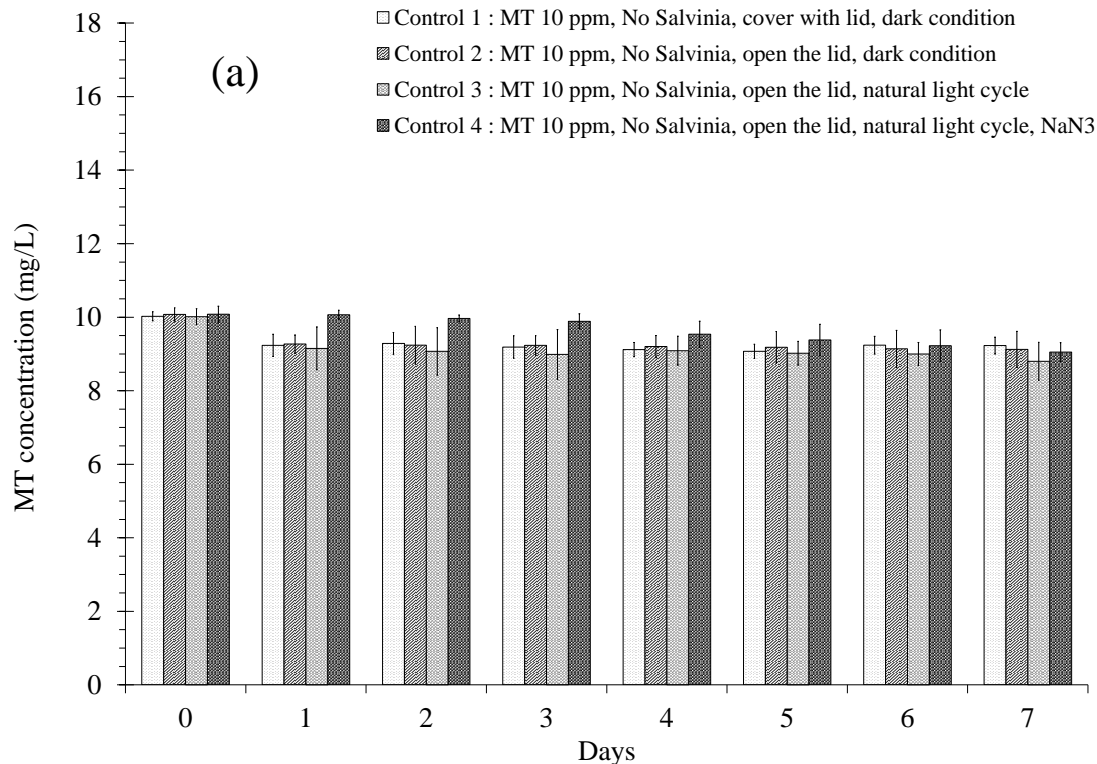
<sup>b</sup> homogenized Salvinia

<sup>c</sup> the sidewalls of the reactor were covered with foil, and the reactor was placed in a dark closed cabinet

# RESULTS AND DISCUSSION

## 1. MT removal in the reactors

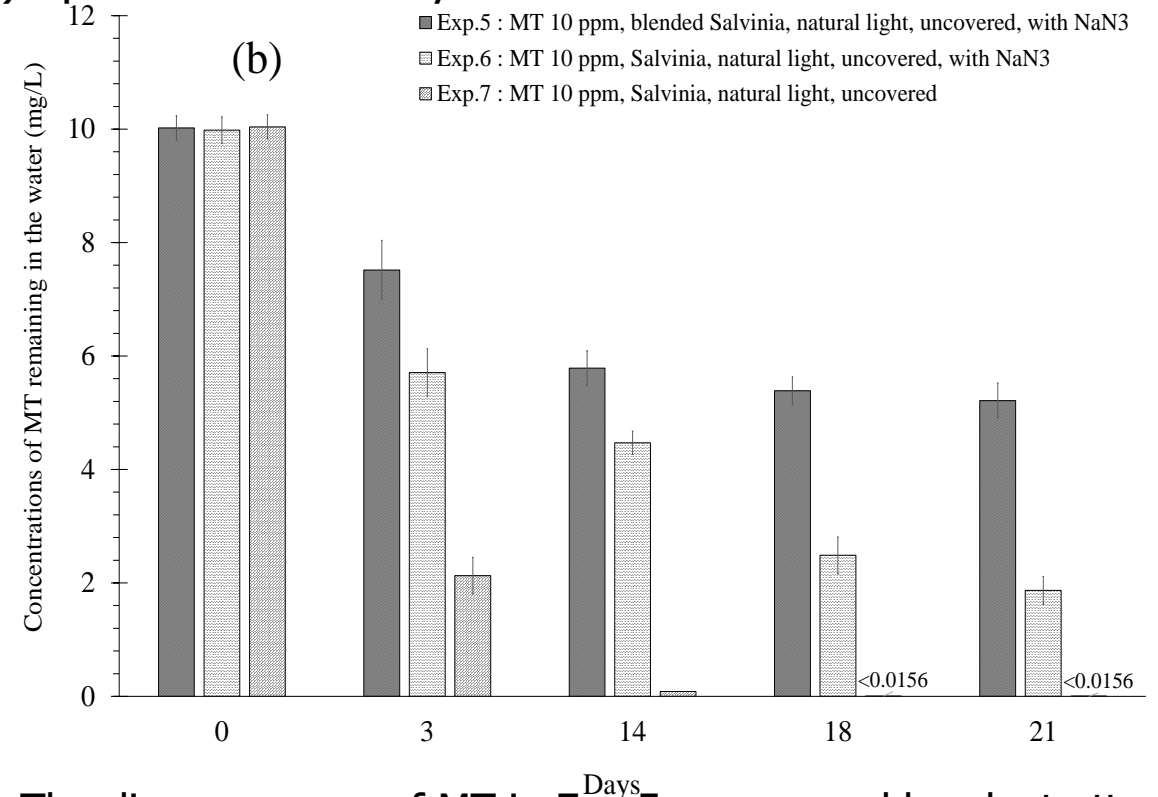
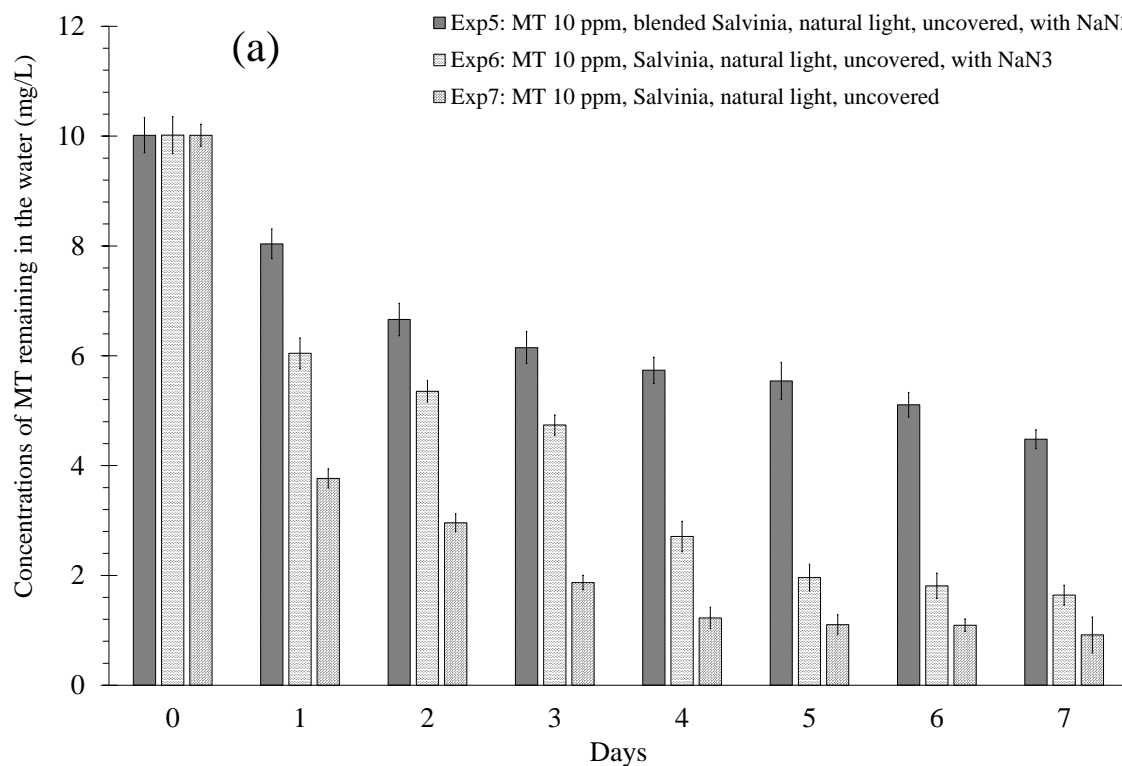
Figure 2 MT concentrations remaining in the reactors in each control experiment for; (a) a retention time of 7 days or (b) a retention time of 21 days.



The loss of MT in controls 1, 2 and 4 was related to physicochemical activities such as the attachment of MT to the glass, volatilization, and photodegradation, while microbial activity resulting from post-contamination from the air might have been the reason for the loss of MT (12.6%) observed in the control 4 reactor.

# RESULTS AND DISCUSSION

Figure 3 MT concentrations remaining in the reactors in each experimental trial; (a) operated for 7 days or (b) operated for 21 days.



- The highest percentage of MT removal observed in the active Salvinia reactor (Exp.7) might have been affected by photodegradation, live plant activities, and microbial activities.
- In Exp.6, a slightly lower value than that in Exp.7 was probably caused by the inhibition of microbial activity.

- The disappearance of MT in Exp.5 was caused by plant-attached microbial communities and the sorption of MT onto Salvinia biomass.
- Increasing the retention time would improve the sorption and transformation of MT via the integrated activities of live plant and microbial communities.

# RESULTS AND DISCUSSION

Table 3 & Table 4 show the MT degradation rates on day 7 and day 21 and half-lives in the reactors

Control/experimental trials	Degradation rate (k) at day-7, (day <sup>-1</sup> )	Degradation rate (k') at day-21, (day <sup>-1</sup> )	Half-life of MT, (day)
Control 1 - No Salvinia + covered top of the reactor + dark conditions	0.000	0.000	-
Control 2 - No Salvinia + dark conditions	0.000	0.000	-
Control 3 - No Salvinia + natural light	0.000	0.000	-
Control 4 - No Salvinia + natural light + NaN <sub>3</sub>	0.001	0.000	-
Exp.5 - Blended Salvinia + natural light + NaN <sub>3</sub>	0.004	0.013	No microbial activity
Exp.6 - Salvinia + natural light + NaN <sub>3</sub>	0.010	0.060	No microbial activity
Exp.7 - Salvinia + natural light	0.014	0.442	1.57

- From Table 3, the biodegradation of MT did not occur within 7 days.
- Increasing the retention time to 21 days would allow the biodegradation process to occur.
- In this study, the biodegradation rate (0.442 day<sup>-1</sup>) at 21 day was higher than found by the previous research (Homklin et al., 2009).

Medium	Conditions	Half-life of E2	Half-life of E1	Half-life of EE2	Half-life of MT
Water	aerobic	0.2-1.7 day <sup>a</sup>	0.2-1.7 day <sup>a</sup>	17 day	1.57 day (This study)

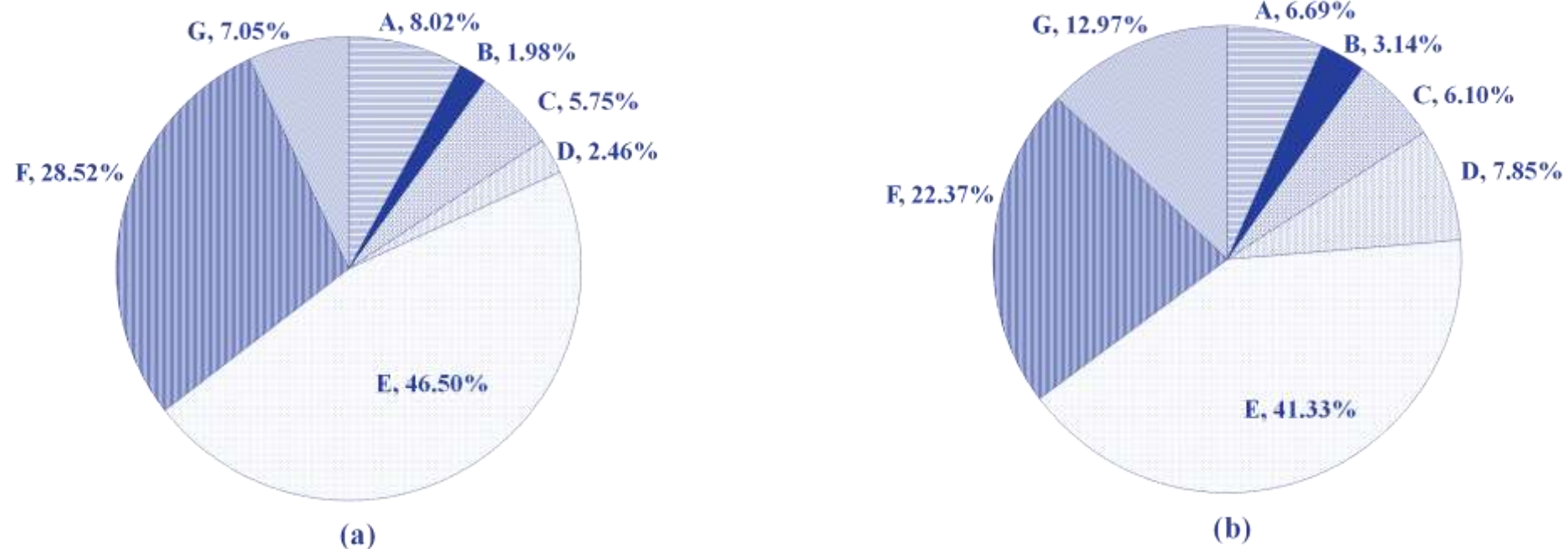
- The half-life of MT fell within the same range as those of E1 and E2.
- This can be explained by the higher similarity of the structure and physicochemical properties of MT to E1 and E2 than to EE2.

Table 5 Contributions of MT depletion in the reactors at retention times of 7 and 21 days.

Contributions to MT removal from the water phase	Calculation of the MT mass fraction	% contribution Day-7	% contribution Day-21
(1) Sorption on glassware	$\text{Mass}_{\text{initial MT}} - \text{Mass}_{\text{control 1}}$	8.02±2.76	6.69±1.89
(2) Volatilization	$\text{Mass}_{\text{control 2}} - \text{Mass}_{\text{control 1}}$	1.98±1.97	3.14±1.78
(3) Photodegradation	$\text{Mass}_{\text{control 3}} - \text{Mass}_{\text{control 2}}$	5.75±1.86	6.10±1.36
(4) Microbial activity (both that originally in the water and post-contamination)	$\text{Mass}_{\text{control 4}} - \text{Mass}_{\text{control 3}}$	2.46±1.09	7.85±1.12
(5) Sorption on plant biomass and plant-attached microbial activity	$\text{Mass}_{\text{control 4}} - \text{Mass}_{\text{Exp.6}}$	46.50±3.80	41.33±2.73
(6) Intact live plant activity (i.e., plant uptake, sorption on live plant roots and attached microbial activity)	$\text{Mass}_{\text{Exp.6}} - \text{Mass}_{\text{Exp.5}}$	28.52±1.19	22.37±1.25
(7) Role of suspended microbial activity	$\text{Mass}_{\text{Exp.7}} - \text{Mass}_{\text{Exp.6}}$	7.05±3.60	12.97±2.84
Total	(1)+(2)+(3)+(4)+(5)+(6)+(7)	100.29±0.48	100.45±0.52

- The role of live plant (*Salvinia*) and overall microbial activities was determined by taking the mass of MT partitioned in the water between control 4 and Exp.7 into account. The contributions (%) on day 7 and day 21 were approximately 82.76% and 82.24%, respectively.
- The mass contribution was rechecked by the summation of the MT mass removed from (5) + (6) + (7).
- The result for the retention time after 7 days was 82.07%, which was very close to the experimental data (82.76%).
- Therefore, it can be concluded that the symbiotic action of microbial activities and *Salvinia* plays a key role in the depletion of MT in the active reactor.

Figure 4 Contribution fractions to MT depletion in the active *Salvinia*-based reactor;  
 (a) retention times of 7 days and (b) 21 days.



**A** = Glassware, **B** = Volatilization, **C** = Photodegradation, **D** = Microbial activity (both that originally in the water and post-contamination), **E** = Sorption on plant biomass and plant-attached microbial communities, **F** = Intact live plant activity (i.e., plant uptake, sorption on live plant roots and attached microbial activity), **G** = Role of suspended microbial activity

To determine the major contributions (%) to MT depletion, a target contribution percentage of 75% was established.

- For a retention time of 7 days, the major contributors were sorption on plant biomass and attached microbial activity (46.50%) and intact live plant activity (28.52%).
- For a retention time of 21 days, there were three major contributors: sorption on plant biomass and assigned microbial activity (41.33%), intact live plant activities (22.37%), and suspended microbial activity (12.97%).
- These findings indicated that an increasing retention time has a significant effect on the increase in suspended microbial activity.



# CONCLUSION

- *Salvinia* can be used as a phytoremediator for the removal of MT from contaminated water.
- Active *Salvinia* reactors can effectively remove MT from aqueous mixtures.
- As a moderately hydrophobic organic contaminant, MT is likely to be taken up and translocated inside plant tissues.
- For a retention time of seven days, the top contributors to the depletion of MT in the active *Salvinia* reactor are sorption on plants and intact live plant activities
- while for a retention time of 21 days major contributors to the depletion of MT in the active *Salvinia* reactor are sorption on plant biomass and assigned microbial activity, intact live plant activities, and suspended microbial activity.
- Microbial activities in the reactor can be enhanced by increasing the retention time.
- Increasing the retention time to 21 days seems to allow sufficient time for the growth of aerobic microbial communities.
- The biodegradation rate and half-life values obtained in this study revealed that aerobic microbial communities are likely to be a key player in the biodegradation of MT in the active *Salvia* reactor.
- The symbiotic activity between intact live plant activities and microbial activities can enhance the biodegradation rate and shorten the half-life in the removal of MT.

Thank you for your kind attention